Biomolecules

Molecules we need to survive

Where did Biomolecules come from?

 To begin learning about biomolecules we must understand how biomolecules were formed.

Scientists knew that early earth had raw materials such as:

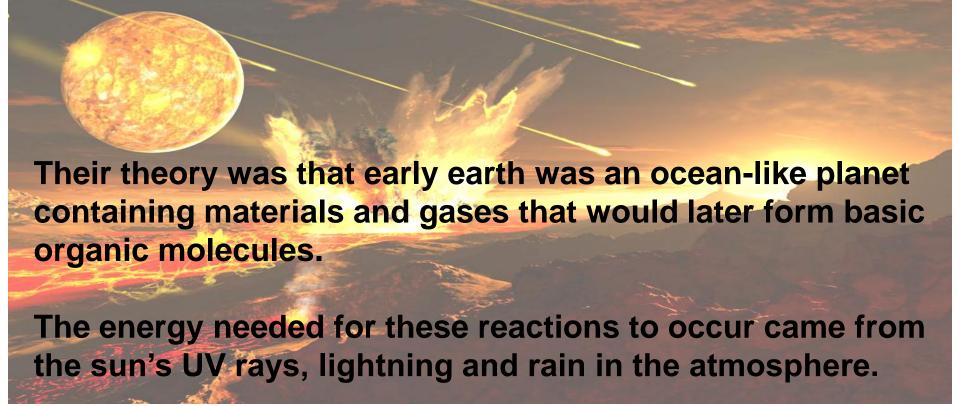
Water,
Hydrogen gas,
Nitrogen,
Ammonia,
Methane,



Carbon dioxide and Carbon monoxide.

1920 - The Primordial Soup Theory is born.

In the early 1920's Russian Chemist A.I. Oparin and English Geneticist J.B.S. Haldane conceived the idea of primordial soup.

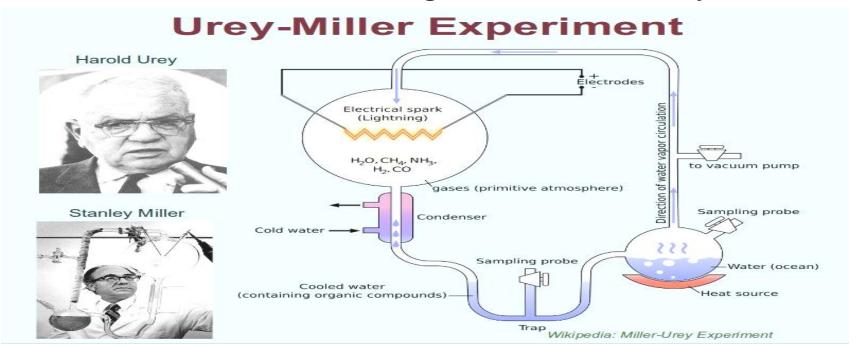


Miller and Urey

- In 1953 Stanley Miller and Harold Urey tested the Primordial Soup theory.
- Click the link below will illustrate their experiment.

http://highered.mcgraw-hill.com/sites/9834092339/student_view0/chapter26/animation_-miller-urey_experiment.html

Miller and Urey showed it was possible to form simple organic molecules from the inorganic material on early Earth.



Molecules covered in this presentation

• 1. Carbohydrates – Sugars "ose"

• 5. Calories

• **6.** ATP

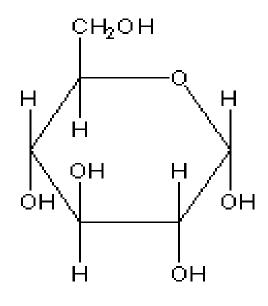
• 2. Lipids – Fats

• **7.** Water

- 3. Proteins (enzymes)
- 8. Vitamins and Minerals

• 4. Nucleic Acids

1. Carbohydrates



C-6 H-12 O-6

Carbohydrates_are the main source of **energy**

Made of C, H, and O atoms in a 1:2:1 ratio

2 types of carbohydrates:

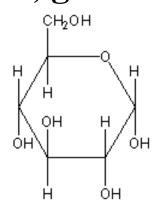
1. - Simple
 Carbohydrates
 (sugars)="ose"

 2. - Complex carbohydrates (starch)

- Have <u>smaller</u> structures and are broken down <u>faster</u> in your body.
- Much <u>larger</u>, take <u>longer</u> to break down.

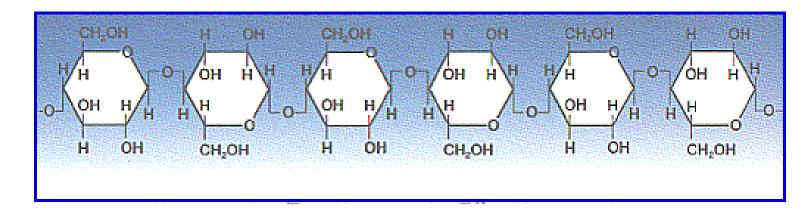
- Ex: Candy gives you a quick boost.
- Ex: pasta, spaghetti, oatmeal, whole grains

Simple sugars are: mono-saccharides (like glucose, fructose, Ribose, galactose)



and <u>di-</u>saccharides (like sucrose, maltose, lactose)

Complex starches are Poly-saccarides

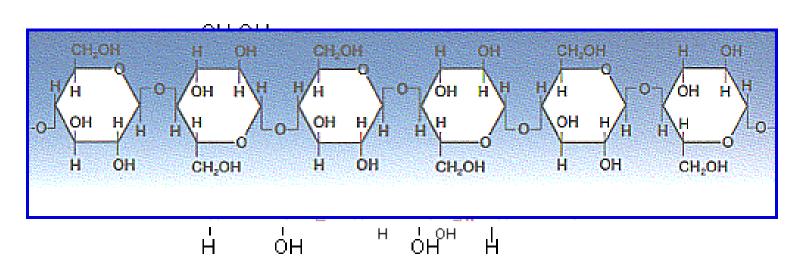


Complex carbohydrates

or Simple Carbohydrate?

Saccharide = sugar

Are these mono-saccharide di-saccharide or poly-saccharide?



Simple

Foods high in simple carbs are fruits, milk, candy, deserts, white flour

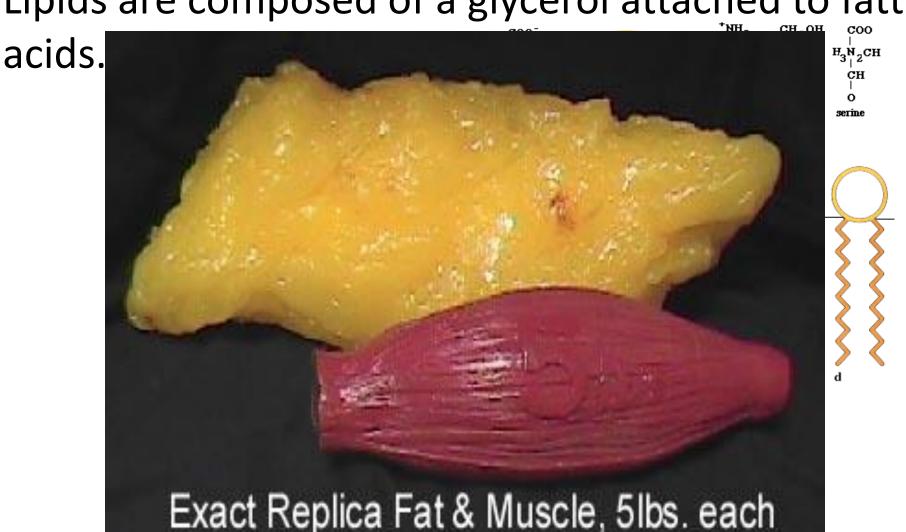






2. Lipids/Fats are mostly C and H atoms and do not dissolve in water.

Lipids are composed of a glycerol attached to fatty



2. Fats (Lipids)

Lipids are used for:

Protect parts of our body

(cushion)



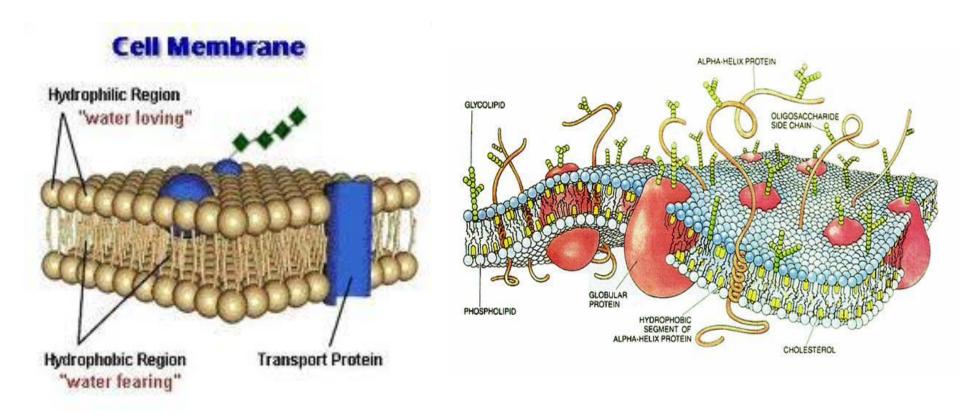


2. Fats (Lipids)

CNS regulation Satiety signals



Lipids are also the main component of the cell membrane!



The cell membrane is also called the – phospholipid bilayer!

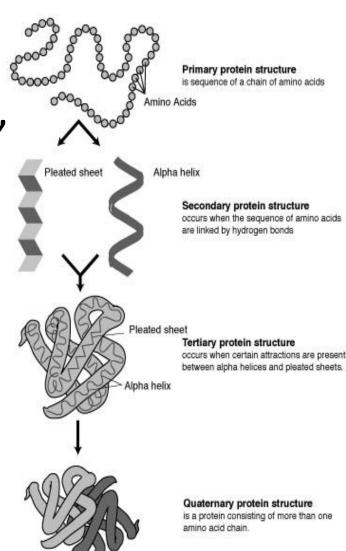


3. Protein

 Make body parts like muscle, bones, hair, etc

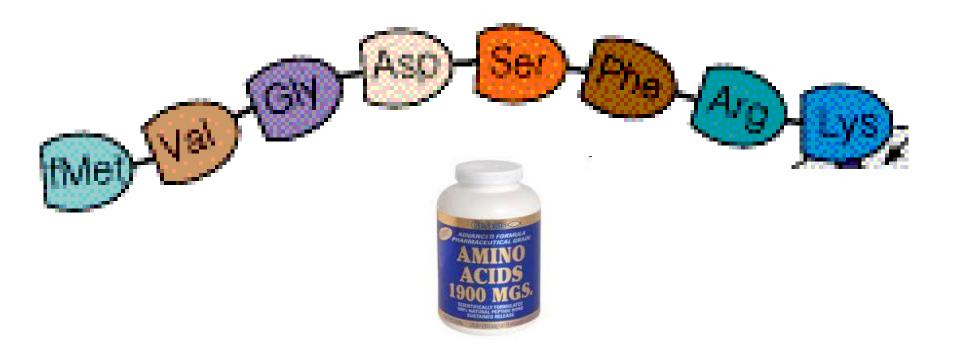
Help control cell functions.

 Can be used for energy (but not the main source)



3. Protein

'•When you put amino acids together in chains you form protein





3. Enzymes "ase" = special proteins that control reactions and cell processes

For example:

Sucrase helps break down

Lactase helps break down

Lipase breaks down (fats)

DNA polymerase helps make

Think of an enzyme like your friend at a party, he helps you hook up faster. He **brings you together** with people you like **faster** than if you went alone.

Enzymes help make reactions happen without using as much energy.

AKA: Lowering the Activation Energy

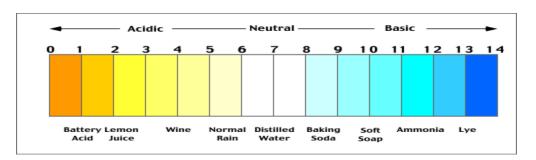
No enzyme Enzyme Energy Substrate **Products**

This increases the <u>speed</u> of the reaction by bringing the substrates together faster

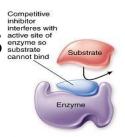
Things that affect the way an Enzyme works

Temperature

•pH



•Inhibitors (blockers or Activators activators or Activato





(a) Competitive inhibition

(b) Noncompetitive inhibition

•Enzyme Cofactors— chemical components that affect how an enzyme works

4. Nucleic Acids

Cytosine

Guanine

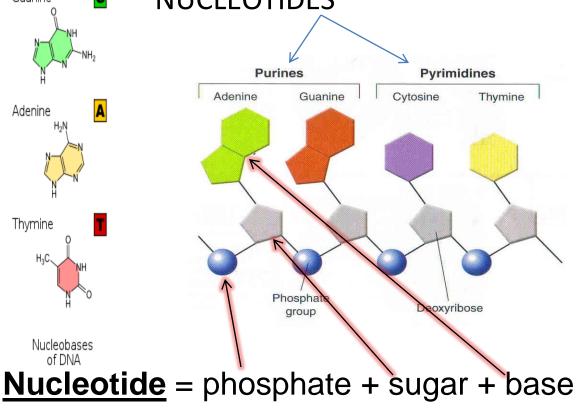
Adenine

Thymine

Nucleobases

Cytosine Nucleobases Guanine Base pair Adenine Uracil helix of sugar-phosphates Nucleobases of RNA RNA DNA Ribonucleic acid Deoxyribonucleic acid Nucleic Acid can be DNA or RNA (NA=Nucleic acid)

Nucleic acids are made up of smaller pieces called **NUCLEOTIDES**



Nutrition Facts Valeur nutritive Per 1 cup (34 g) / pour 1 tasse (34 g) % Dally Val Amount Teneur % valeur "Joudienne Calories / Calories 120 Fat/Lipides 3 g 5% Saturated / saturés 1 g 10% + Trans / trans 1 g Cholesterol / Cholestérol 0 mg Sodium / Sodium 220 mg 9% Potassium / Potassium 55 mg 2 % Carbohydrate / Glucides 27 g 9% Fibre / Fibres 1 g 3 % Sugars / Sucres 14 g Protein / Protéines 1 q 10% Vitamin A / Vitamine A Vitamin C / Vitamine C 10% Calcium / Calcium 5 % Iron / Fer 20% Vitamin D / Vitamine D 10% Thiamine / Thiamine 20 % Riboflaven / Riboflavine 20 % Niacin / Niacine 20 % Vitamin B6 / Vitamine B6 20 % Folic Acid / Acide folique 20 % Vitamin B12 / Vitamine B12 20 % Zinc / Zinc 10%

5. Calories

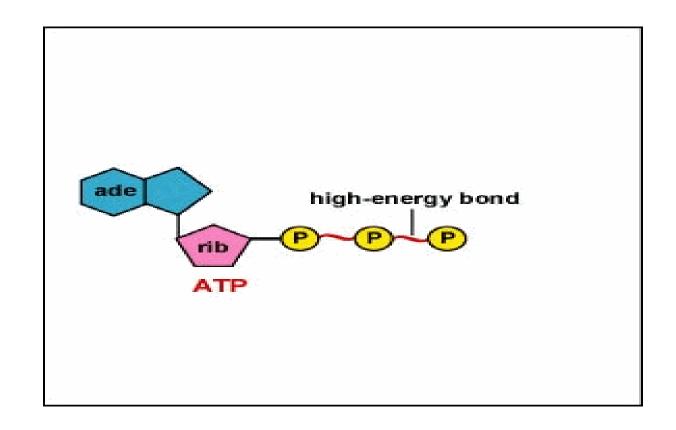
- Calories are units of energy.
- They measure the amount of energy in the food we eat

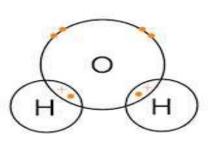
 Carbohydrates, Proteins, and Fats all have calories (Fats have the most)

 The average person needs around 2000 Calories a day (If you eat more than that your body may store the rest as fat)

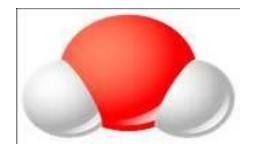
6. ATP Adenosine TriPhosphate

 Our food is changed into energy our cells can use called <u>ATP</u>



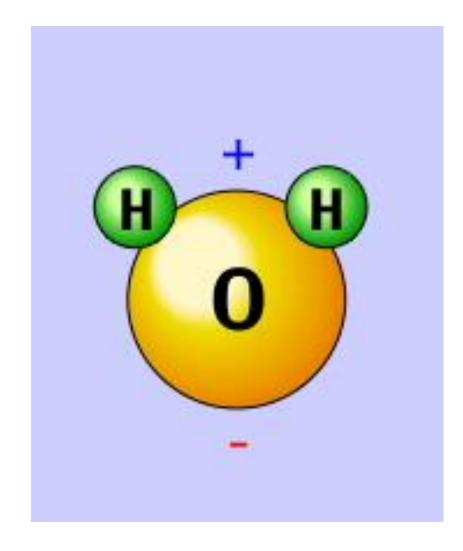


7. Water



 Water is the universal solvent (it can dissolve almost anything)

 Water is polar
 Water has a positive + and a negative side –



Your body needs water!! In fact around 60% of your body is water and over 70% of your brain!

You can survive without food longer than without water.

8. Vitamins and Minerals

- Vitamins and Minerals are also necessary for your body (but in small amounts)
- You can get all of these by eating a well balanced diet!!!!

